

Synthesis of imbiline 1, a tetracyclic aza-aromatic alkaloid

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Abstract

A tetracyclic aza-aromatic alkaloid, imbiline 1 (**1**) was synthesised from 1-amino-4-methoxynaphthalene hydrochloride (**7**) in seven steps. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: alkaloids; naphthyridines; oxidation; cyclisation.

A number of polycyclic aromatic alkaloids have been isolated from natural resources.¹ Imbilines 1 (**1**), 2 (**2**), 3 (**3**) and eupomatidines 1 (**4**), 2 (**5**), 3 (**6**) are tetracyclic aromatic alkaloids isolated from *Eupomatia bennettii* and *E. laurina* (Fig. 1).² We achieved total synthesis of **4–6** possessing iminoquinolinequinone structure.³ Now we report the synthesis of **1**.

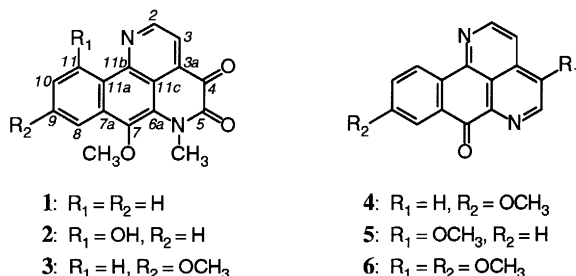
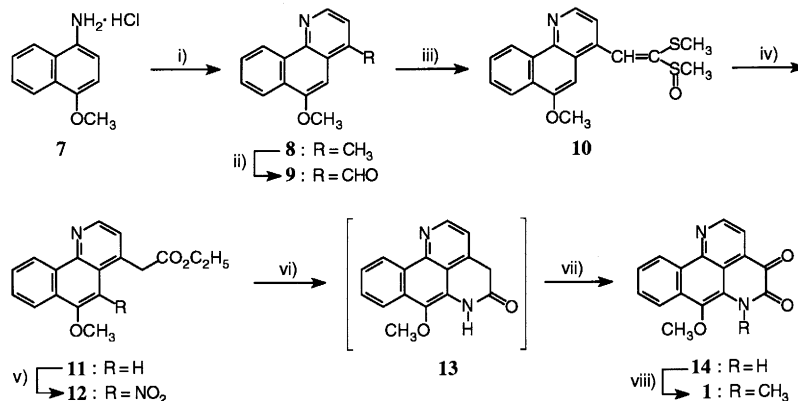


Fig. 1.

1-Amino-4-methoxynaphthalene hydrochloride⁴ (**7**) was treated with methyl vinyl ketone under acidic conditions to give benzo[*h*]quinoline (**8**) in 75% yield (Scheme 1). Oxidation of **8** using Vismara's method⁵ afforded aldehyde (**9**) in 84% yield. The aldehyde (**9**) was treated with methyl methylsulfinylmethyl sulfide and sodium hydroxide followed by hydrochloric acid in ethanol⁶ to furnish ethyl ester (**11**) in 60% yield.⁷ Nitration of **11** was carried out with potassium nitrate and sulfuric acid–fuming sulfuric acid in nitromethane at $-30^\circ C$ to give 5-nitrobenzo[*h*]quinoline (**12**)⁸ in 65% yield. Treatment of

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12 with sodium hydrosulfite in aqueous tetrahydrofuran under argon formed 4,5-dihydronaphtho[1,2,3-*ij*][2,7]naphthyridin-5(6*H*)-one (**13**). We found that the benzylic position of the tetracyclic compound (**13**) was easily oxidised under air and a keto amide (**14**) was isolated in 62% yield from **12**. Finally **14** was methylated with methyl iodide in dimethylsulfoxide to furnish imbiline 1 (**1**) in 46% yield. The spectral data of synthetic compound **1**⁹ and imbiline 1 obtained from natural resources were identical. The present method should be generally applicable for the synthesis of imbilines 2 (**2**) and 3 (**3**) possessing 4,5-dihydronaphtho[1,2,3-*ij*][2,7]naphthyridine-4,5(6*H*)-dione skeleton.



Scheme 1. (i) Methyl vinyl ketone, HCl, ZnCl₂, *m*-nitrobenzenesulfonic acid, ethanol, reflux, 3 h; (ii) trifluoroacetic acid, *tert*-butyl iodide, iodine, FeCl₂·4H₂O, DMSO, 90°C, 12 h; (iii) methyl methylsulfinylmethyl sulfide, NaOH, 60°C, 1 h; (iv) HCl, ethanol, reflux, 3 h; (v) KNO₃, concentrated H₂SO₄–fuming H₂SO₄ (4:1), nitromethane, –30°C, 10 min; (vi) Na₂S₂O₄, tetrahydrofuran, H₂O, 25°C, 1 h; (vii) air oxidation; (viii) methyl iodide, KOH, DMSO, 25°C, 1 h.

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7. An attempted reaction of **8** with *n*-butyllithium in tetrahydrofuran followed by ethyl chloroformate failed, giving 2-(*n*-butyl)-6-methoxy-4-methylbenzo[*h*]quinoline in 76% yield.
8. Ethyl (6-methoxy-5-nitrobenzo[*h*]quinolin-4-yl)acetate (**12**): mp 129–131°C (yellow needles from ethanol). MS *m/z* (%): 340 (M⁺, 1.6), 295 (27), 294 (100), 266 (84), 238 (22). Anal. calcd for C₁₈H₁₆N₂O₅: C, 63.53; H, 4.74; N, 8.23. Found: C, 63.73; H, 4.81; N, 8.14. IR (KBr): 1730, 1530, 1372 cm^{–1}. ¹H NMR (270 MHz, CDCl₃) δ: 1.30 (3H, t, *J*=7.6 Hz, CH₂CH₃), 3.93 (2H, s, CH₂CO), 4.12 (3H, s, OCH₃), 4.22 (2H, q, *J*=7.6 Hz, CH₂CH₃), 7.43 (1H, d, *J*=4.6 Hz, C₃-H), 7.80–7.89 (2H, m, C₈-H, C₉-H), 8.17–8.20 (1H, m, C₇-H), 8.94 (1H, d, *J*=4.6 Hz, C₂-H), 9.38–9.42 (1H, m, C₁₀-H).
9. 7-Methoxy-6-methyl-4,5-dihydronaphtho[1,2,3-*ij*][2,7]naphthyridine-4,5(6*H*)-dione (imbiline 1 (**1**)): mp 237–240°C (red prisms from ethanol) [lit.² mp 212–214°C]. MS *m/z* (%): 292 (M⁺, 98), 249 (100). High-resolution MS calcd for C₁₇H₁₂N₂O₃: 292.0848. Found: 292.0847. IR (KBr): 1694, 1670 cm^{–1}. ¹H NMR (270 MHz, CDCl₃) δ: 3.97 (3H, s, OCH₃), 4.02 (3H, s, NCH₃), 7.78 (1H, td, *J*=7.3, 1.7 Hz, C₁₀-H), 7.83 (1H, td, *J*=7.3, 1.7 Hz, C₉-H), 8.24 (1H, dd, *J*=7.3, 1.7 Hz, C₈-H), 8.27 (1H, d, *J*=4.6 Hz, C₃-H), 9.21 (1H, dd, *J*=7.3, 1.7 Hz, C₁₁-H), 9.22 (1H, d, *J*=4.6 Hz, C₂-H). ¹³C NMR (67.5 MHz, CDCl₃) δ: 35.87 (NCH₃), 62.97 (OCH₃), 118.02 (C_{11c}), 119.26 (C₃), 122.64 (C_{6a}), 122.73 (C₈), 125.05 (C₁₁), 128.30 (C₁₀), 130.17 (C₉), 130.33 (C_{11a}), 130.55 (C_{7a}), 132.00 (C_{3a}), 144.69 (C₇), 145.50 (C_{11b}), 148.64 (C₂), 157.72 (C₅), 176.68 (C₄).